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Journal

English LA 63-5 (Pharmaceuticals) CC Inclusion complexes of prednisolone (PR) with .beta.-cyclodextrin AB (.beta.-CD) and hydroxypropyl-.beta.-cyclodextrin (HP.beta.-CD) were formed by the solvation method, and were characterized by DSC, x-ray diffractometry and FT-IR spectroscopy. PC liposomes incorporating PR as plain drug or inclusion complex were prepd. using the dehydrationrehydration method and drug entrapment as well as drug release were estd. for all liposome types prepd. The highest PR entrapment value (80% of the starting material) was achieved for PC/Chol liposomes when the HP.beta.-CD-PR (2:1, mol/mol) complex was entrapped. The leakage of vesicle encapsulated 5,6-carboxyfluorescein (CF) was used as a measure of the vesicle membrane integrity. As judged from our exptl. results liposomes which encapsulate .beta.-CD-PR complexes are significantly less stable (when their membrane integrity is considered) compared to liposomes of identical lipid compns. which incorporate plain drug or even (in some cases) non-drug incorporating liposomes, which were prepd. and studied for comparison. Interestingly, liposomes which encapsulate HP.beta.-CD-PR complexes, have very low initial CF latency values, indicating that the leakage of CF is a process of very high initial velocity. Interactions between lipid and cyclodextrin mols. may be possibly resulting in rapid reorganization of the lipid membrane with simultaneous fast release of CF mols. The release of PR from liposomes was highest when the drug was entrapped in the form of a complex with .beta.-CD. Nevertheless, the very high entrapment ability of PR in the form of HP.beta.-CD-PR complexes in comparison to plain drug is a indubitable advantage of this approach. prednisolone cyclodextrin complex liposome ST ITDissolution rate Encapsulation (liposomes encapsulating prednisolone and prednisolone-cyclodextrin complexes) Inclusion compounds IT RL: PRP (Properties); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (liposomes encapsulating prednisolone and prednisolone-cyclodextrin -complexes) Drug delivery systems IT(liposomes; liposomes encapsulating prednisolone and prednisolone-cyclodextrin complexes) 50-24-8, Prednisolone ITRL: PRP (Properties); RCT (Reactant); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (liposomes encapsulating prednisolone and prednisolone-cyclodextrin complexes) 57-55-6D, 1,2-Propanediol, ether with .beta.-cyclodextrin, complex with ITprednisolone 7585-39-9D, .beta.-Cyclodextrin, hydroxypropyl ether, complex with prednisolone 370094-76-1 RL: PRP (Properties); THU (Therapeutic use); BIOL (Biological study); USES (Uses) (liposomes encapsulating prednisolone and prednisolone-cyclodextrin complexes) RE.CNT THERE ARE 28 CITED REFERENCES AVAILABLE FOR THIS RECORD 28 RE(1) Arimori, K; J Inclus Phenom 1984, V1, P387 CA (2) Arimori, K; J Pharm Dyn 1987, V10, P390 CA (3) Castelli, F; Int J Pharm 1992, V88, P1 CA (4) Cleland, M; Biochim Biophys Acta 1981, V597(2), P418 (5) Debouzy, J; J Pharm Sci 1998, V87, P59 CA (6) Fatouros, D; J Drug Targeting 2001, V9(1), P61 CA (7) Fauvelle, F; J Pharm Sci 1997, V86, P935 CA (8) Frank, S; J Pharm Sci 1975, V64, P1585 CA (9) Fukuda, N; Chem Pharm Bull 1986, V34, P1366 CA (10) Kirby, C; Biochem Pharmacol 1983, V32, P609 CA (11) Kirby, C; Biotechnology 1984, V2, P979 CA

(12) Kokkona, M; Eur J Pharm Sci 2000, V9, P245 CA (13) Loftsson, T; J Pharm Sci 1996, V85, P1017 CA (14) McCormack, B; Biochim Biophys Acta 1996, V1291, P237 CA (15) McCormack, B; Int J Pharm 1994, V112, P249 CA (16) McCormack, B; J Drug Targeting 1994, V2, P449 CA (17) New, R; Liposomes: A Practical Approach 1990 (18) Nishijo, J; Chem Pharm Bull 1998, V46(1), P120 CA (19) Nishijo, J; Chem Pharm Bull 2000, V48(1), P48 CA (20) Saenger, W; Angew Chem Int Ed 1980, V19, P344 (21) Senior, J; Life Sci 1982, V30, P2123 CA (22) Stewart, J; Anal Biochem 1980, V104, P10 CA (23) Takino, T; Biol Pharm Bull 1994, V17, P121 CA (24) Uekama, K; CRC Crit Rev Ther Drug Carrier Systems 1987, V3, P1 CA (25) Uekama, K; Int J Pharm 1982, V10, P1 CA (26) Uekama, K; J Pharm Dyn 1983, V6, P124 CA (27) Uekama, K; Pharm Helv Acta 1985, V60, P117 CA (28) Weeks, C; J Am Chem Soc 1973, V95, P2865 CA L2ANSWER 2 OF 12 CA COPYRIGHT 2002 ACS AN 135:269016 CA The effect of arbutin on membrane integrity during drying is mediated by TI stabilization of the lamellar phase in the presence of nonbilayer-forming lipids Oliver, A. E.; Hincha, D. K.; Tsvetkova, N. M.; Vigh, L.; Crowe, J. H. AU Section of Molecular and Cellular Biology, University of California, CS Davis, CA, 95616, USA Chem. Phys. Lipids (2001), 111(1), 37-57 SO CODEN: CPLIA4; ISSN: 0009-3084 Elsevier Science Ireland Ltd. PBJournal DTEnglish LA6-6 (General Biochemistry) CC Section cross-reference(s): 11 Arbutin (4-hydroxyphenyl-.beta.-glucopyranoside) is a solute accumulated AB to high concns. in drought and frost resistant plants. Arbutin can inhibit membrane lysis, both free radical-mediated and enzymic in nature, and it has been suggested that arbutin might contribute to membrane stabilization in these plants. However, we found that arbutin destabilized phosphatidylcholine vesicles during drying and rehydration, which appears to be inconsistent with the proposed protective function of arbutin for membranes. We also found, however, that arbutin stabilizes membranes contg. nonbilayer-forming lipids during freezing. We now report that, in liposomes contq. the nonbilayer-forming lipids monogalactosyldiacylglycerol (MGDG) or phosphatidylethanolamine (PE), arbutin served a protective function during drying, as measured by retention of carboxyfluorescein (CF) and extent of vesicle fusion. In hydrated samples contg. these lipids, arbutin stabilized the lamellar liq. cryst. phase. Therefore, the interaction between arbutin and lipid membranes and the resulting effects on membrane stability depend, in a complex manner, on the lipid compn. of the membrane. arbutin membrane integrity lamellar phase lipid stabilization dessication ST tolerance ITOsmolality (arbutin effect on membrane integrity during drying in relation to) Liposomes ITMembrane, biological (arbutin effect on membrane integrity during drying is mediated by stabilization of lamellar phase in presence of nonbilayer-forming lipids) Lipids, biological studies IT Phosphatidylcholines, biological studies Phosphatidylethanolamines, biological studies RL: BSU (Biological study, unclassified); PEP (Physical, engineering or chemical process); BIOL (Biological study); PROC (Process)

(arbutin effect on membrane integrity during drying is mediated by stabilization of lamellar phase in presence of nonbilayer-forming lipids) Stress, plant (dessication; arbutin effect on membrane integrity during drying is mediated by stabilization of lamellar phase in presence of nonbilayer-forming lipids) Membrane phase transition, biological (hexagonal II-lamellar; arbutin effect on membrane integrity during drying is mediated by stabilization of lamellar phase in presence of nonbilayer-forming lipids) Diglycerides RL: BSU (Biological study, unclassified); PEP (Physical, engineering or chemical process); BIOL (Biological study); PROC (Process) (monogalactosyl; arbutin effect on membrane integrity during drying is mediated by stabilization of lamellar phase in presence of nonbilayer-forming lipids) Organelle (vesicle; arbutin effect on membrane integrity during drying is mediated by stabilization of lamellar phase in presence of nonbilayer-forming lipids) 57-50-1, SUCROSE, biological studies 99-20-7, Trehalose 7447-40-7, Potassium chloride, biological studies 9005-27-0, Hydroxyethyl starch RL: BAC (Biological activity or effector, except adverse); BIOL (Biological study) (effect on membrane integrity during drying) 497-76-7, Arbutin RL: BAC (Biological activity or effector, except adverse); BIOL (Biological study) (effect on membrane integrity during drying is mediated by stabilization of lamellar phase in presence of nonbilayer-forming lipids) THERE ARE 67 CITED REFERENCES AVAILABLE FOR THIS RECORD RE.CNT 67 (1) Appelqvist, L; J Lipid Res 1972, V13, P146 CA (2) Aurell Wistrom, C; Biochim Biophys Acta 1989, V984, P238 -(3) Bianchi, G; Physiol Plant 1993, V87, P223 CA (4) Bligh, E; Can J Biochem Physiol 1959, V37, P911 CA (5) Bryszewska, M; Biochim Biophys Acta 1988, V943, P485 CA (6) Buitink, J; Plant Physiol 1996, V111, P235 CA (7) Crowe, J; Ann Rev Physiol 1992, V54, P79 (8) Crowe, J; Ann Rev Physiol 1998, V60, P73 CA (9) Crowe, J; Arch Biochem Biophys 1984, V232, P400 CA (10) Crowe, J; Biochim Biophys Acta 1988, V939, P327 CA (11) Crowe, J; Cryobiology 1994, V31, P355 MEDLINE (12) Crowe, J; Cryobiology 1997, V35, P20 CA (13) Crowe, J; Science 1984, V223, P701 CA (14) Crowe, L; Arch Biochem Biophys 1985, V242, P240 CA (15) Crowe, L; Biophys J 1996, V71, P2087 CA (16) Cullis, P; Biochim Biophys Acta 1978, V507, P207 CA (17) Cullis, P; Chem Phys Lipids 1986, V40, P127 CA (18) de Kruijff, B; Nature 1987, V329, P587 MEDLINE (19) Drennan, P; J Plant Physiol 1993, V142, P493 CA (20) Golovina, E; Plant Physiol 1998, V118, P975 CA (21) Gounaris, K; Biochim Biophys Acta 1983, V732, P229 CA (22) Green, J; J Phys Chem 1989, V93, P2880 CA (23) Hincha, D; Biophys J 1999, V77, P2024 CA (24) Hinz, H; Biochemistry 1991, V30, P5125 CA (25) Hoekstra, F; Comp Biochem Physiol 1997, V117A, P335 CA (26) Horvath, G; Biochim Biophys Acta 1987, V891, P67 (27) Horvath, I; Proc Natl Acad Sci USA 1998, V95, P3513 CA (28) Ingram, J; Ann Rev Plant Physiol Plant Mol Biol 1996, V47, P377 CA (29) Ioku, K; Biosci Biotech Biochem 1992, V56, P1658 CA (30) Janes, N; Chem Phys Lipids 1996, V81, P133 CA (31) Lafleur, M; Biophys J 1996, V70, P2747 CA

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- L2 ANSWER 3 OF 12 CA COPYRIGHT 2002 ACS
- AN 131:98984 CA
- TI Stabilizing effect of an s-layer on liposomes towards thermal or mechanical stress
- AU Mader, C.; Kupcu, S.; Sara, M.; Sleytr, U. B.
- CS Zentrum fur Ultrastrukturforschung und Ludwig Boltzmann-Institut fur Molekulare Nanotechnologie, Universitat fur Bodenkultur Wien, Vienna, A-1180, Austria
- SO Biochim. Biophys. Acta (1999), 1418(1), 106-116 CODEN: BBACAQ; ISSN: 0006-3002
- PB Elsevier Science B.V.
- DT Journal
- LA English
- CC 6-3 (General Biochemistry)
- AB Isolated subunits of the cryst. cell surface layer (S-layer) protein of Bacillus stearothermophilus PV72/p2 were recrystd. on pos. charged unilamellar liposomes. Liposomes were composed of dipalmitoylphosphatidylcholine (DPPC), cholesterol and hexadecylamine (HDA) in a molar ratio of 10:5:4 and they were prepd. by the dehydration-rehydration method followed by an extrusion procedure. The S-layer protein to DPPC ratio was 5.7 nmol/.mu.mol which approx. corresponds to the theor. value estd. by using the areas occupied by the S-layer lattice and the lipid membrane. Coating of the pos. charged liposomes with S-layer protein resulted in inversion of the .zeta.-potential from +29.1 mV to -27.1 mV. Covalent crosslinking of the recrystd. S-layer protein was achieved with glutaraldehyde. Chem. anal. revealed that almost all amino groups (> 95%) from HDA in the liposomal membrane were involved in the reaction. To study the influence

of an S-layer lattice on the stability of the liposomes, the hydrophilic marker carboxyfluoresceine (CF) was encapsulated and its release was detd. for plain and S-layer-coated liposomes in the course of mech. and thermal challenges. In comparison to plain liposomes, S-layer-coated liposomes released only half the amt. of enclosed CF upon exposure to shear forces or ultrasonication as mech. stress factors. Furthermore, temp. shifts from 25.degree. to 55.degree. and vice versa induced considerably less CF release from S-layer-coated than from plain liposomes. A similar stabilizing effect of the S-layer lattice was obsd. after glutaraldehyde treatment of plain and S-layer-coated liposomes.

ST S layer protein liposome stabilization

IT Proteins, specific or class

RL: BPR (Biological process); BIOL (Biological study); PROC (Process) (S-layer (surface layer); stabilizing effect of S-layer on liposomes towards thermal or mech. stress)

IT Electric potential

(biol.; stabilizing effect of S-layer on liposomes towards thermal or mech. stress)

IT Stability

(stabilizing effect of S-layer on liposomes towards thermal or mech. stress)

IT Liposomes

(unilamellar; stabilizing effect of S-layer on liposomes towards thermal or mech. stress)

IT 57-88-5, Cholesterol, biological studies 63-89-8,

Dipalmitoylphosphatidylcholine 143-27-1, 1-Hexadecanamine

RL: BPR (Biological process); BIOL (Biological study); PROC (Process) (stabilizing effect of S-layer on liposomes towards thermal or mech. stress)

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(33) Tilcock, C; Biochim Biophys Acta 1989, V979, P208 CA (34) Weigert, S; J Membr Sci 1995, V106, P147 CA (35) Wetzer, B; Langmuir 1998, V14, P6899 CA ANSWER 4 OF 12 CA COPYRIGHT 2002 ACS L2124:5774 CA AN A chloride channel from human placenta reconstituted into giant liposomes TI Riquelme, Gloria; Stutzin, Andres; Barros, Luis Felipe; Liberona, Jose AU Luis Facultad de Medicina, Universidad de Chile, Santiago, 7, Chile CS Am. J. Obstet. Gynecol. (1995), 173(3, Pt. 1), 733-8 SO CODEN: AJOGAH; ISSN: 0002-9378 Journal DTEnglish LA13-6 (Mammalian Biochemistry) CC Ion channels play important roles in epithelial transport, but they are AB difficult to access for conventional electrophysiol. studies in intact placenta. The purpose of this work was to explore the suitability of purified trophoblast plasma membrane as a source of ion channels for reconstitution in artificial lipid membranes. Human placental brush border membranes were purified by differential and gradient centrifugation and fused with small liposomes. Giant liposomes were then generated by a cycle of dehydration and rehydration. These giant liposomes are suitable for electrophysiol. studies and were probed for the presence of active ion channels by the patch-clamp method. The results reported here indicate the presence of a high conductance chloride channel showing some similarities with "maxi" chloride channels described in secreting and absorbing epithelia. The channel had a slight outward rectification with conductances of 232 and 300 pS at neg. and pos. potentials, resp. For the first time successful reconstitution of a human placental ion channel is achieved in a system suited for electrophysiol. studies. The chloride channel described might play a role in transplacental transport. chloride channel placenta reconstitution liposome STBrush border IT(brush border membrane in reconstitution of human placental ion channel -in giant liposomes) ITLiposome Placenta (reconstitution of human placental ion channel in giant liposomes) IT Cell membrane Trophoblast (trophoblast plasma membrane in reconstitution of human placental ion channel in giant liposomes) Ion channel IT(chloride, reconstitution of human placental ion channel in giant liposomes) 16887-00-6, Chloride, biological studies ${ t IT}$ RL: BPR (Biological process); BIOL (Biological study); PROC (Process) (reconstitution of human placental ion channel in giant liposomes) L2ANSWER 5 OF 12 CA COPYRIGHT 2002 ACS AN 119:23204 CA TI The effects of beryllium on the electrostatic and thermodynamic properties of the dipalmitoyllecithin membranes Ermakov, Yu. A.; Mahmudova, S. S.; Shevchenko, E. V.; Lobyshev, V. I. ΑU Frumkin A. N., Inst. Electrochem., Moscow, Russia CS SO Biol. Membr. (1993), 10(2), 212-24 CODEN: BIMEE9; ISSN: 0233-4755 DTJournal Russian LACC 6-6 (General Biochemistry) Section cross-reference(s): 4 AB The boundary potential at the lipid membrane /electrolyte interface consists of two components: surface potential

.phi.s and dipole component .phi.d. The first one is included in the widely used Gouy-Chapman-Stern formalism to explain the electrostatic phenomena, the second one is sensitive to the orientation effects of lipid polar groups and water mols. Both components were studied with a combination of electrophoretic measurements in the liposome suspension, sensitive to .phi.s and by the method of the inner membranous field compensation in the planar lipid membranes sensitive to changes in the sum .phi.s + .phi.d. It is shown that both potentials and the thermotropic properties of the DPPC membranes depend on the divalent cation adsorption and beryllium is the most effective cation. The boundary potential components .phi.s and .phi.d are changed in opposite directions at the lipid phase transition temps. A new pool of lipids proportional to surface area occupied by cations is obsd. by calorimetry and has the phase transition temp. depending on cation surface concn. The comparison of the data obtained in the electrolytes prepd. with normal water and D2O reveals strong isotope effects. It is concluded that the cation adsorption is accompanied by rehydration processes and structural modifications of the lipid membrane surface. beryllium dipalmitoylphosphatidylcholine bilayer potential phase transition Phosphatidylcholines, biological studies RL: BIOL (Biological study) (bilayer membrane, electrostatic and thermotropic properties of, beryllium effect on) Heat of transition (gel-liq. cryst., of dipalmitoylphosphatidylcholine bilayer, beryllium effect on) Hydration, chemical (of dipalmitoylphosphatidylcholine bilayer membrane, beryllium effect on) Isotope effect (on dipalmitoylphosphatidylcholine bilayer membrane phase transition in beryllium presence, of deuterium) Membrane, biological (bilayer, dipalmitoylphosphatidylcholine, electrostatic and thermotropic properties of, beryllium effect on) Cations (divalent, dipalmitoylphosphatidylcholine bilayer membrane electrostatic and thermotropic properties response to) Membrane phase transition, biological (gel-liq. cryst., of dipalmitoylphosphatidylcholine bilayer, beryllium effect on) Electric activity (potential, dipolar, of dipalmitoylphosphatidylcholine bilayer membrane, in beryllium presence, phase transition effect on) Electric activity (potential, surface, of dipalmitoylphosphatidylcholine bilayer membrane, in beryllium presence, phase transition effect on) Membrane phases, biological (sepn., of dipalmitoylphosphatidylcholine bilayer, beryllium induction of) 63-89-8 RL: BIOL (Biological study) (bilayer membrane, electrostatic and thermotropic properties of, beryllium effect on) 7440-41-7, Beryllium, biological studies RL: BIOL (Biological study) (dipalmitoylphosphatidylcholine bilayer membrane electrostatic and thermotropic properties response to) 7782-39-0, Deuterium, biological studies RL: PRP (Properties) (isotope effect of, on dipalmitoylphosphatidylcholine bilayer membrane

phase transition in beryllium presence)

ST

IT

ANSWER 6 OF 12 CA COPYRIGHT 2002 ACS L2 AN 118:229767 CA Roles of water molecules in bacteria and viruses TICox, C. S. AU Chem. Biol. Def. Establ., Porton Down/Salisbury, SP4 0JQ, UK CS Origins Life Evol. Biosphere (1993), 23(1), 29-36 SO CODEN: OLEBEM; ISSN: 0169-6149 Journal; General Review DT English LA 10-0 (Microbial, Algal, and Fungal Biochemistry) CC A review with 3 refs. In addn. to water, microbes mainly comprise lipids, AB carbohydrates, proteins, and nucleic acids. Their structure and function singularly and conjointly are affected by water activity. Desiccation leads to dramatic lipid phase changes, whereas carbohydrates, proteins, and nucleic acids initially suffer spontaneous, reversible low activation energy Maillard reactions forming products that more slowly rearrange, crosslink, etc., to give nonnative states. While initial products spontaneously may reverse to native states by raising water activity, later products only do so through energy consumption and enzymic activity (e.g., repair). Yet, native states of lipid membranes and assocd. enzymes are required to generate energy. Consequently, good reserves of high energy compds. (e.g., ATP) and of membrane stabilizers (e.g., trehalose) may be expected to enhance survival following drying and rehydration (e.g., anhydrobiotic organisms). ŞT review water bacteria virus IT Bacteria Virus (water in, roles of) 7732-18-5, Water, biological studies IT RL: BIOL (Biological study) (in bacteria and viruses, roles of) L2ANSWER 7 OF 12 CA COPYRIGHT 2002 ACS 118:35221 CA AN TIThe cryoprotective mechanism of saccharides on freezing and freeze-drying of liposomes Miyajima, Koichiro; Tanaka, Keiko AU Fac. Pharm. Sci., Kyoto Univ., Kyoto, 606, Japan Trends Glycosci. Glycotechnol. (1992), 4(19), 457-63 SO CODEN: TGGLEE; ISSN: 0915-7352 DTJournal; General Review English LA9-0 (Biochemical Methods) CC Section cross-reference(s): 6 A review with 9 refs. about studying the title mechanism by several AB methods, such as leakage of aq. inner marker, Raman- and NMR-spectroscopy, The surface of frozen liposome was covered by a concd. aq. saccharide soln. or glassy solid, which protects from the mech. damage of ice crystal and the fusion of liposome. Mono-, di-, and trisaccharides showed a similar protective effect per monosaccharide unit. In the course of drying, the water mols. hydrated to the polar phosphate group of lecithin were displaced by the saccharide mols. In the liq. crystal state the lyophilized liposome was maintained and the lipid membrane was stable during the rehydration process. The liposome lyophilized with a disaccharide showed the strongest stability during the rehydration process, indicating the importance of hydrogen bonding between the phosphate group and a sugar mol. with suitable mol. size. ST review liposome freeze drying saccharide; cryoprotectant liposome saccharide review Monosaccharides IT Oligosaccharides RL: PRP (Properties) (cryoprotective mechanism of, on freezing and freeze drying of liposomes)

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IT
     Liposome
        (freezing and freeze drying of, cryoprotective mechanism of saccharides
        on)
     Freeze drying
IT
     Freezing
        (of liposomes, cryoprotective mechanism of saccharides in)
     Cryoprotectants
IT
        (saccharides, in freezing and freeze drying of liposome)
     ANSWER 8 OF 12 CA COPYRIGHT 2002 ACS
L2
AN
     110:179517 CA
     Liver-targeted pharmaceutical liposomes containing iminoacetic
TI
     acid-chromium-hepatobiliary targeting agent coordination compounds
IN
     Geho, W. Blair; Lau, John R.
     USA
PA
     PCT Int. Appl., 48 pp.
SO
     CODEN: PIXXD2
     Patent
DT
     English
LA
IC
     ICM A61K049-00
     63-6 (Pharmaceuticals)
CC
     Section cross-reference(s): 1
FAN.CNT 1
                                           APPLICATION NO.
     PATENT NO.
                      KIND
                            DATE
                                                            DATE
PI
                            19880128
     WO 8800474
                                           WO 1986-US1421
                       A1
                                                            19860710
         W: AU, BR, DK, FI, GB, JP, KR, LK, NL, NO, SE
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     AU 8661413
                            19880210
                       A1
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                                                            19860710
     AU 607682
                       B2
                            19910314
     EP 274467
                       A1
                            19880720
                                           EP 1986-904629
                                                            19860710
                       B1
     EP 274467
                            19920520
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                                           JP 1986-503988
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     DK 8801256
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                       Α
     DK 169502
                       B1
                            19941114
     NO 8801058
                            19880510
                                           NO 1988-1058
                       Α
                                                            19880309
                            19860710
PRAI WO 1986-US1421
     Pharmaceuticals comprise a diagnostic agent, a lipid
AB
     membrane structure in form of a vesicle or liposome and a
     component comprising a fatty substituent attached to the vesicle wall and
     a target-seeking substituent selected from a class of chems. that have a
     high affinity to hepatobiliary receptors. A delivery system comprising a
     serotonin-contg. lipid vesicle, a connector [N-(2,6-
     diisopropylphenylcarbamoylmethyl)iminoacetic acid, Hepatolite], a bridge
     (Cr), and a hepatobiliary target-seeking agent (biliveridin) was prepd. A
     lipid film contg. distearoyl lecithin 69.12, iminodiacetic acid complex
     (Hepatolite) 1.07, dicetyl phosphate 14.1, and cholesterol 5.0 mg was
     rehydrated with a mixt. contg. phosphate buffer, human serum
     albumin, and serotonin. Unencapsulated serotonin and albumins were
     removed and the vesicles were treated with a 5-fold molar excess of CrCl3,
     excess CrCl3 was subsequently removed, and the vesicles were treated with
     a 5-fold molar excess of connector mols., and finally, excess connector
     mols. were also removed. The vesicles thus prepd. were connected to
     biliverdin to give a hepatocyte-directed drug delivery vesicle. A dog in
     a state of net hepatic glucose output was infused with 0.3 .mu.g/kg per
     min serotonin-charged hepatocyte-directed vesicle; the hepatic glucose
     output was converted to hepatic uptake and the uptake was maintained for
     30 min after the serotonin infusion was discontinued. Liver storage of
     glucose eaten during meal requires not only insulin but also serotonin.
    Hepatolite itself can also be used as hepatobiliary target-seeking agent.
    liver targeted liposome drug bioavailability
ST
    Receptors
IT
    RL: BIOL (Biological study)
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(hepatobiliary, coordination compds. with iminodiacetic acid and chromium, liver-targeted liposomes contg.) Drug bioavailability (in liver, targeted liposomes for, contg. coordination compds. of chromium and iminodiacetic acid and hepatobiliary targeting agents) Albumins, biological studies Hormones RL: BIOL (Biological study) (liver-targeted pharmaceutical liposomes contg. iminodiacetic acid-chromium-hepatobiliary targeting agent coordination compds. and) Liver (targeted liposomes for, contg. coordination compds. of chromium and iminodiacetic acid and hepatobiliary receptors) Diabetes mellitus (treatment of, liver-targeted pharmaceutical insulin-contg. liposomes for) Liver (hepatocyte, targeted drug delivery to, with liposomes contg. iminodiacetic acid-chromium-hepatobiliary receptor targeting agent coordination compds.) Pharmaceutical dosage forms (liposomes, liver-targeted, contg. iminodiacetic acid-chromiumhepatobiliary targeting agent coordination compds.) Diabetes mellitus (maturity-onset, treatment of, liver-targeted pharmaceutical insulin-contg. liposomes for) 114-25-0D, chromium complexes 142-73-4D, Iminodiacetic acid, complexes with chromium and hepatobiliary target-seeking agents 7440-47-3D, Chromium, hepatolite-biliverdin complex 120093-62-1D, complexes with chromium and hepatobiliary targeting agents RL: BIOL (Biological study) (liver-targeted pharmaceutical liposomes contg.) 50-67-9, Serotonin, biological studies 57-88-5, Cholesterol, biological 2197-63-9, Dicetyl phosphate studies 4539-70-2, Distearoyl lecithin 9004-10-8, Insulin, biological studies 9002-72-6, Growth hormone RL: BIOL (Biological study) (liver-targeted pharmaceutical liposomes contg. iminodiacetic acid-chromium-hepatobiliary targeting agent complexes and) 7440-47-3D, Chromium, complexes with hepatobiliary targeting agents and iminodiacetate RL: BIOL (Biological study) (liver-targeted pharmaceutical liposomes contg. pharmaceuticals) ANSWER 9 OF 12 CA COPYRIGHT 2002 ACS 106:143922 CA Lyophilized liposomes prepared by a modified reversed-phase evaporation method Handa, Tetsurou; Takeuchi, Hirofumi; Ohokubo, Yuichi; Kawashima, Yoshiaki Gifu Pharm. Univ., Gifu, 502, Japan Chem. Pharm. Bull. (1987), 35(2), 748-55 CODEN: CPBTAL; ISSN: 0009-2363 Journal English 63-6 (Pharmaceuticals) A modification of the reversed-phase evapn. method (modified REV method) was developed for the prepn. of lyophilized unilamellar liposomes. The encapsulation efficiencies of the liposomes after a dehydration (lyophilization) - rehydration procedure were satisfactorily high, and the liposome sizes were maintained nearly const. throughout the procedure. These results are different from those obtained with liposome samples prepd. by the reversed-phase evapn. method. In the latter case,

marked enlargement of liposome size and extensive leakage from liposomes

keeps the lipid membranes fluid even at freezing temp.

were obsd. The small amt. of residual ether in the modified REV liposomes

The fluidity is considered to play an important role in the protection of

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liposomes against aggregation, fusion and leakage.
     lyophilization liposome reversed phase evapn
ST
     Phosphatidylcholines, biological studies
IT
     RL: BIOL (Biological study)
        (liposomes contg., prepn. of lyophilized, by modified reversed-phase
        evapn.)
     Evaporation
IT
        (liposomes prepd. by modified reversed-phase)
IT
     Liposome
        (lyophilized, modified reversed phase evapn. in prepn. of)
IT
     Freeze drying
        (of liposomes prepd. by modified reversed phase evapn.)
     Pharmaceutical dosage forms
IT
        (liposomes, prepn. of lyophilized, by modified reversed phase evapn.)
     57-88-5, Cholesterol, biological studies 124-30-1, Stearylamine
IT
     3614-36-6, Diacetyl phosphate
     RL: BIOL (Biological study)
        (liposomes contg., prepn. of lyophilized, by modified reversed-phase
        evapn.)
     ANSWER 10 OF 12 CA COPYRIGHT 2002 ACS
L2
     97:19414 CA
AN
     Encapsulation of macromolecules by lipid vesicles under simulated
TI
     prebiotic conditions
     Deamer, David W.; Barchfeld, Gail L.
AU
     Univ. California, Davis, CA, 95616, USA
CS
     J. Mol. Evol. (1982), 18(3), 203-6
SO
     CODEN: JMEVAU; ISSN: 0022-2844
     Journal
DT
     English
LA
CC
     6-6 (General Biochemistry)
     Phospholipid vesicles (liposomes) were subjected to dehydration-hydration
AB
     cycles in the presence of 6-carboxyfluorescein or salmon sperm DNA.
     vesicles fused into multilamellar structures during dehydration with
     solutes trapped between the lamellae. On rehydration the
     lamellae swelled and formed large vesicular structures contg. solute.
     This model can be used to study encapsulation of macromols. by
     lipid membranes to form protocellular structures under
     prebiotic conditions.
     phospholipid liposome encapsulation DNA carboxylfluorescein; evolution
ST
     phospholipid liposome encapsulation macromol
     Hydration, chemical
IT
        (-dehydration, of phosphatidylcholine liposome in macromol.
        encapsulation, evolution in relation to)
    Dehydration, chemical
IT
        (-hydration, of phosphatidylcholine liposome in macromol.
        encapsulation, evolution in relation to)
     Phosphatidylcholines, biological studies
IT
     RL: BIOL (Biological study)
        (liposome, macromol. encapsulation by, evolution in relation to)
IT
     Encapsulation
        (of macromols., by phosphatidylcholine liposomes, evolution in relation
        to)
    Deoxyribonucleic acids
IT
     RL: BIOL (Biological study)
        (phosphatidylcholine liposome encapsulation of, evolution in relation
        to)
IT
    Liposome
        (phosphatidylcholine, macromol. encapsulation by, evolution in relation
        to)
    Evolution
IT
        (prebiotic, phosphatidylcholine liposome encapsulation of
        carboxyfluorescein or DNA in relation to)
IT
    3301-79-9
    RL: BIOL (Biological study)
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(phosphatidylcholine liposome encapsulation of, evolution in relation to)

ANSWER 11 OF 12 CA COPYRIGHT 2002 ACS L294:98380 CA ANProtein-lipid interactions in biological and model membrane systems. TIDeuterium NMR of Acholeplasma laidlawii B, Escherichia coli, and cytochrome oxidase systems containing specifically deuterated lipids Kang, Shyue-Yue; Kinsey, Robert A.; Rajan, Srinivasan; Gutowsky, Herbert AU S.; Gabridge, Michael G.; Oldfield, Eric Dep. Chem., Univ. Illinois, Urbana, IL, 61801, USA CS J. Biol. Chem. (1981), 256(3), 1155-9 SO CODEN: JBCHA3; ISSN: 0021-9258 Journal DTEnglish LA6-13 (General Biochemistry) CC 2H NMR spectra of A. laidlawii B (PG9) membranes and lipid exts. enriched ABbiosynthetically in the presence of avidin, with either [14-2H3]tetradecan-1-oic acid, [16-2H3]hexadecan-1-oic acid, [4-2H2]-, [6-2H2]-, or [8-2H2]-tetradecan-1-oic acids, were recorded at a variety of temps. At their growth temp. the A. laidlawii membrane lipids are .apprx.90% in a rigid gel-like state. Plasma membranes which had been lyophilized, then rehydrated, behaved in the 2H NMR expt. as did fresh plasma membranes. The 2H NMR quadrupole splittings (.DELTA..nu.Q) were very similar for all of the fluid phase spectra recorded. These results indicate that protein has little effect on lipid order in the A. laidlawii B membrane system. The 2H-quadrupole splittings obsd. for the tetradecanoic acid-enriched membranes were within exptl. error the same as those obsd. previously for bilayers of pure 1,2-myristoyl-sn-glycero-3phosphocholine (DMPC) when examd. immediately above the end of the solid-to-fluid phase transition temp. range. Relatively small decreases in order in the DMPC mol. were seen using cytochrome oxidase as a model membrane protein at high protein to lipid ratio, the effects being largest near the chain terminus (C12-C14). By contrast, 2H NMR spectra of the hexadecan-1-oic acid-enriched Escherichia coli L48-2 cell membranes showed extreme line broadening compared to spectra of their lipid exts., and .DELTA..nu.Q values were slightly decreased. Results with intact E. coli cell membranes show essentially the same NMR line shapes as those seen previously with the DMPC-gramidicin A' system including collapsed terminal Me group quadrupole splittings and large (4-6 kHz) line widths of methylene segment chain resonances. cytochrome oxidase lipid membrane; Acholeplasma STmembrane lipid protein; Escherichia membrane lipid protein; protein lipid interaction membrane NMR Proteins ITRL: BIOL (Biological study) (lipid interactions with, in plasma membranes of Acholeplasma and Escherichia) Acholeplasma laidlawii ITEscherichia coli (lipid-protein interactions in plasma membrane of) IT Cell membrane (lipid-protein interactions in, of Acholeplasma and Escherichia) Nuclear magnetic resonance IT(of deuterium, of plasma membranes of Acholeplasma and Escherichia) Order IT(of plasma membrane lipids of Acholeplasma and Escherichia) Lipids IT RL: BIOL (Biological study) (protein interactions with, in plasma membranes of Acholeplasma and Escherichia) 18194-24-6 ITRL: BIOL (Biological study) (cytochrome oxidase interaction with, in model membranes) 9001-16-5 IT

RL: BIOL (Biological study) (lecithin interaction with, in model membranes) 57-10-3, biological studies 544-63-8, biological studies IT RL: BIOL (Biological study) (Acholeplasma plasma membranes enriched in, lipid-protein interactions in) ANSWER 12 OF 12 CA COPYRIGHT 2002 ACS L2AN 92:112612 CA Improving and storage stability of aqueous dispersions of spherules by TIlyophilization Vanlerberghe, Guy; Handjani, Rose Marie INL-Oreal, Fr. PABrit. UK Pat. Appl., 7 pp. SO CODEN: BAXXDU Patent \mathbf{DT} English LĄ B01J013-00 IC CC 45-4 (Fats and Waxes) Section cross-reference(s): 62, 63 FAN.CNT 1 PATENT NO. APPLICATION NO. KIND DATE DATE PIGB 2013609 19790815 GB 1979-3477 19790201 A GB 2013609 19821208 B2 FR 1978-2927 FR 2416008 19790831 19780202 A1 FR 2416008 B1 19810626 ES 477351 A1 19791216 ES 1979-477351 19790131 US 4247411 19810127 US 1979-8115 Α 19790131 BE 873865 19790801 BE 1979-193210 A1 19790201 NL 7900822 19790806 NL 1979-822 A 19790201 DE 2904047 19790913 DE 1979-2904047 A1 19790202 DE 2904047 C2 19900419 C2 DE 2954558 19900517 19790202 DE 1979-2954558 PRAI FR 1978-2927 19780202 Aq. dispersions of 100-50,000-.ANG.-diam. liposomes with an encapsulated aq. phase contg. an active substance, e.g. a cosmetic or pharmaceutical, were stabilized for storage by lyophilization and reconstituted by Thus, C16H33[OCH2CH(CH2OH)]nOH [72920-85-5] (av. n rehydration. = 3) 190, cholesterol [57-88-5] 190, and Na dicetyl phosphate [60285-46-3] 20 mg were mixed at 90.degree., cooled to 70.degree., and homogenized with 2.5 mL 2% aq. Na L-pyrrolidonecarboxylate (I) [28874-51-3]. The homogenized mixt. was ultrasonically dispersed 30 min in a further 7.5 mL 2% aq. I to give a fluid dispersion contg. .apprx.1-.mu.-diam. liposomes. The dispersion was cooled in liq. N and lyophilized 12 h to form a storage-stable, white, pasty product which on rehydration with >3.5 mL H2O gave liposomes of the same diam. as the original ones. liposome stabilization lyophilization; encapsulation cosmetic liposome ST storage; pharmaceutical encapsulation liposome storage; lipid membrane spherule storage Lecithins, compounds ITRL: USES (Uses) (hydrogenated, liposomes contg., stabilization of, lyophilization in) Membranes and Diaphragms IT(lipid, stabilization of, lyophilization in) ITCosmetics Pharmaceuticals (liposome-encapsulated, stabilization of, lyophilization in) ITFreeze drying (of liposomes, stabilization by) ITLiposome (stabilization of, lyophilization in relation to) ITImmunoglobulins RL: USES (Uses)

(A, liposomes contg., stabilization of, lyophilization in) Alcohols, compounds IT RL: USES (Uses) (lanolin, hydrogenated, ethers with oligomeric poly(glycerol ether), liposomes contg., stabilization of, by lyophilization) 50-81-7, uses and miscellaneous 56-81-5, uses and miscellaneous IT57-88-5, uses and miscellaneous 72-17-3 9054-89-1 28874-51-3 60285-46-3 72920-21-9 RL: USES (Uses) (liposomes contg., stabilization of, lyophilization in) 72920-88-8D, monoethers, with hydrogenated lanolin alcs. IT72920-85-5

RL: USES (Uses) (oligomeric, liposomes contg., stabilization of, lyophilization in)

=> b biosis

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FILE COVERS 1969 TO DATE.

CAS REGISTRY NUMBERS AND CHEMICAL NAMES (CNs) PRESENT FROM JANUARY 1969 TO DATE.

RECORDS LAST ADDED: 6 March 2002 (20020306/ED)

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4668 REHYDRAT?

193563 LIPID

772819 MEMBRANE?

4697 LIPID(W) MEMBRANE?

L3 8 L1 AND LIPID(W) MEMBRANE?

=> s 13 not 12

4668 REHYDRAT?

193563 LIPID

772819 MEMBRANE?

4697 LIPID(W) MEMBRANE?

L4 0 L3 NOT L2

=> b medline

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3789 REHYDRAT?

138368 LIPID

568357 MEMBRANE?

2604 LIPID(W) MEMBRANE?

6 L1 AND LIPID(W) MEMBRANE? L5

=> s 15 not 12

3789 REHYDRAT?

138368 LIPID

568357 MEMBRANE?

2604 LIPID(W) MEMBRANE?

L6

0 L5 NOT L2

=> b uspatfull

COST IN U.S. DOLLARS SINCE FILE

TOTAL **SESSION** ENTRY

FULL ESTIMATED COST

36.52 0.32

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)

SINCE FILE

ENTRY

TOTAL SESSION

CA SUBSCRIBER PRICE

-7.08 0.00

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FILE COVERS 1971 TO PATENT PUBLICATION DATE: 7 Mar 2002 (20020307/PD) FILE LAST UPDATED: 7 Mar 2002 (20020307/ED)

HIGHEST GRANTED PATENT NUMBER: US6353930

HIGHEST APPLICATION PUBLICATION NUMBER: US2002029398

CA INDEXING IS CURRENT THROUGH 7 Mar 2002 (20020307/UPCA)

ISSUE CLASS FIELDS (/INCL) CURRENT THROUGH: 7 Mar 2002 (20020307/PD)

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- >>> publications, starting in 2001, for the inventions covered in
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- published document but also a list of any subsequent
- publications. The publication number, patent kind code, and
- >>> publication date for all the US publications for an invention <<<
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- records and may be searched in standard search fields, e.g., /PN, <<< >>> /PK, etc. >>>
- >>> USPATFULL and USPAT2 can be accessed and searched together
- >>> through the new cluster USPATALL. Type FILE USPATALL to

>>> enter this cluster.

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<<< >>> Use USPATALL when searching terms such as patent assignees, <<< >>> classifications, or claims, that may potentially change from <<< the earliest to the latest publication. <<< This file contains CAS Registry Numbers for easy and accurate substance identification. => s 12 5750 REHYDRAT? 28549 LIPID 150714 MEMBRANE? 1248 LIPID(W) MEMBRANE? 137 L1 AND LIPID(W) MEMBRANE? L7 => s 17 and ionophore? 2675 IONOPHORE? 15 L7 AND IONOPHORE? L8=> d ti 1-15 L8 ANSWER 1 OF 15 USPATFULL TI Release of therapeutic agents in a vessel or tissue ANSWER 2 OF 15 L8 USPATFULL Method of producing an electrode membrane combination TIL8ANSWER 3 OF 15 USPATFULL TI Self assembly of sensor membranes L8 ANSWER 4 OF 15 USPATFULL Liposome drug-loading method and composition TI L8 ANSWER 5 OF 15 USPATFULL TIMethod of producing a first layer electrode membrane for a biosensor ANSWER 6 OF 15 USPATFULL L8Methods and apparatus for making liposomes containing hydrophobic drugs ΤI ANSWER 7 OF 15 USPATFULL rsEncapsulation of antineoplastic agents in liposomes TIL8 ANSWER 8 OF 15 USPATFULL Gas and gaseous precursor filled microspheres as topical and TIsubcutaneous delivery vehicles L8 ANSWER 9 OF 15 USPATFULL Methods and apparatus for making liposomes TI L8 ANSWER 10 OF 15 USPATFULL TIFusogenic lipsomes and methods for making and using same ANSWER 11 OF 15 USPATFULL L8 Stable plurilamellar vesicles TI $\mathbf{L8}$ ANSWER 12 OF 15 USPATFULL Encapsulation of antineoplastic agents in liposomes ${f TI}$ L8 ANSWER 13 OF 15 USPATFULL TI Stable plurilamellar vesicles L8ANSWER 14 OF 15 USPATFULL Means of preparation and applications of liposomes containing high TIconcentrations of entrapped ionic species

ANSWER 15 OF 15 USPATFULL $\Gamma8$ Lipid vesicles bearing carbohydrate surfaces as lymphatic directed TI vehicles for therapeutic and diagnostic substances => d cit 2 3 5 'CIT' IS NOT A VALID FORMAT FOR FILE 'USPATFULL' The following are valid formats: The default display format is STD. ABS ----- AB ALL ----- AN, TI, IN, INA, PA, PAA, PAT, PI, AI, PTERM, DCD, RLI, PRAI, DT, FS, REP, REN, EXNAM, LREP, CLMN, ECL, DRWN, AB, GOVI, PARN, SUMM, DRWD, DETD, CLM, INCL, INCLM, INCLS, NCL, NCLM, NCLS, IC, ICM, ICS, EXF, ARTU ALLG ----- ALL plus PAGE.DRAW BIB ----- AN, TI, IN, INA, PA, PAA, PAT, PI, AI, PTERM, DCD, RLI, PRAI, DT, FS, EXNAM, LREP, CLMN, ECL, DRWN, LN.CNT BIB.EX ---- BIB for original and latest publication BIBG ----- BIB plus PAGE.DRAW BROWSE ---- See "HELP BROWSE" or "HELP DISPLAY BROWSE". BROWSE must entered on the same line as DISPLAY, e.g., D BROWSE. CAS ----- OS, CC, SX, ST, IT CBIB ----- AN, TI, IN, INA, PA, PAA, PAT, PI, AI, PRAI, DT, FS DALL ----- ALL, delimited for post-processing FP ----- PI, TI, IN, INA, PA, PAA, PAT, PTERM, DCD, AI, RLI, PRAI, IC, ICM, ICS, INCL, INCLM, INCLS, NCL, NCLM, NCLS, EXF, REP, REN, ARTU, EXNAM, LREP, CLMN, DRWN, AB FP.EX ----- FP for original and latest publication FPALL ----- PI, TI, IN, INA, PA, PAA, PAT, PETRM, DCD, AI, RLI, PRAI, IC, ICM, ICS, INCL, INCLM, INCLS, NCL, NCLM, NCLS, EXF, REP, REN, ARTU, EXNAM, LREP, CLMN, DRWN, AB, PARN, SUMM, DRWD, DETD, CLM FPBIB ----- PI, TI, IN, INA, PA, PAA, PAT, PTERM, DCD, AI, RLI, PRAI, REP, REN, EXNAM, LREP, CLM, CLMN, DRWN FHITSTR ---- HIT RN, its text modification, its CA index name, and its structure diagram FPG ----- FP plus PAGE.DRAW GI ----- PN and page image numbers HIT ----- All fields containing hit terms HITRN ----- HIT RN and its text modification HITSTR ---- HIT RN, its text modification, its CA index name, and its structure diagram IABS ----- ABS, indented with text labels IALL ----- ALL, indented with text labels IALLG ----- IALL plus PAGE.DRAW IBIB ----- BIB, indented with text labels IBIB.EX ---- IBIB for original and latest publication IBIBG ----- IBIB plus PAGE.DRAW IMAX ----- MAX, indented with text labels IMAX.EX ---- IMAX for original and latest publication IND ----- INCL, INCLM, INCLS, NCL, NCLM, NCLS, IC, ICM, ICS, EXF, ARTU, OS, CC, SX, ST, IT ISTD ----- STD, indented with text labels KWIC ----- All hit terms plus 20 words on either side MAX ----- AN, TI, IN, INA, PA, PAA, PAT, PI, AI, PTERM, DCD, RLI, PRAI, DT, FS, REP, REN, EXNAM, LREP, CLMN, ECL, DRWN, AB, GOVI, PARN, SUMM, DRWD, DETD, CLM, INCL, INCLM, INCLS, NCL, NCLM, NCLS, IC, ICM, ICS, EXF, ARTU OS, CC, SX, ST, IT MAX.EX ---- MAX for original and latest publication

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OCC ----- List of display fields containing hit terms
SBIB ----- AN, TI, IN, INA, PA, PAA, PAT, PI, AI, RLI, PRAI,
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SCAN ----- AN, TI, NCL, NCLM, NCLS, IC, ICM, ICS (random display
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       2002:19170 USPATFULL
AN
       Method of producing an electrode membrane combination
TI
       Raguse, Burkhard, St. Ives, AUSTRALIA
IN
       Pace, Ronald John, Homebush, AUSTRALIA
       King, Lionel George, Merefield, AUSTRALIA
       Braach-Maksvytis, Vijoleta Lucija, Dulwich Hill, AUSTRALIA
       Cornell, Bruce, Neutral Bay, AUSTRALIA
       Australian Membrane and Biotechnology Research Institute, Chattsworth,
PA
       AUSTRALIA (non-U.S. corporation)
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                               19990304 (9)
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       US 1999-262097
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RLI
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PRAI
       AU 1995-3669
DT
       Utility
       GRANTED
FS
      Primary Examiner: Chin, Christopher L.
EXNAM
       Gottlieb, Rackman & Reisman, P.C.
LREP
CLMN
       Number of Claims: 22
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       Exemplary Claim: 1
       23 Drawing Figure(s); 15 Drawing Page(s)
DRWN
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CAS INDEXING IS AVAILABLE FOR THIS PATENT.
     ANSWER 3 OF 15 USPATFULL
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AN
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TI
       Self assembly of sensor membranes
       Raguse, Burkhard, St. Ives, Australia
IN
       Pace, Ronald John, Homebush, Australia
       King, Lionel George, Morefield, Australia
       Braach-Maksvytis, Vijoleta Lucija, Dulwich Hill, Australia
       Cornell, Bruce, Neutral Bay, Australia
PA
       Australian Membrane and Biotechnology Research Institute, Chattsworth,
       Australia (non-U.S. corporation)
PΙ
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ΑI
       US 1999-262098
                               19990304 (9)
       Division of Ser. No. US 685329, now patented, Pat. No. US 5879878
RLI
PRAI
       AU 1995-3669
                           19950620
DT
       Utility
FS
       GRANTED
      Primary Examiner: Chin, Christopher L.
EXNAM
LREP
       Gottlieb, Rackman & Reisman, P.C.
CLMN
       Number of Claims: 40
ECL
       Exemplary Claim: 1
DRWN
       23 Drawing Figure(s); 15 Drawing Page(s)
LN.CNT 1003
CAS INDEXING IS AVAILABLE FOR THIS PATENT.
L8
     ANSWER 5 OF 15 USPATFULL
AN
       1999:30564 USPATFULL
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Method of producing a first layer electrode membrane for a biosensor TI Raguse, Burkhard, St. Ives, Australia IN Pace, Ronald John, Homebush, Australia King, Lionel George, Marafield, Australia Braach-Makavytie, Vijoleta Licija, Dulwich Hill, Australia Cornell, Bruce, Neutral Bay, Australia Australian Membrane and Biotechnology Research Institute, Chatswood, PAAustralia (non-U.S. corporation) The University of Sydney, Sydney, Australia (non-U.S. corporation) US 5879878 19990309 PI19960723 (8) ΑI US 1996-685329 PRAI AU 1995-3669 19950620 WO 1996-AU369 19960620 Utility DT Granted FS EXNAM Primary Examiner: Chin, Christopher L. Gottlieb, Rackman & Reisman, P.C. LREP Number of Claims: 15 CLMN Exemplary Claim: 1 ECL 23 Drawing Figure(s); 15 Drawing Page(s) DRWN LN.CNT 920

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